

# SAFETY DATA SHEETS

According to the UN GHS revision 8

Version: 1.0 Creation Date: July 15, 2024 Revision Date: July 15, 2024

1.	Identification			
1.1	GHS Product identifier			
Produ	uct name			
1.2	Other means of identification			
Produ	uct number	C14126		
Other	names			
1.3	Recommended use of the chemical and restrictions on use			
	Identified uses	Processing Aids and Additives		
	Uses advised against	no data available		
1.4	Supplier's details			
	Company	Tianjin Psaitong Biomedical Technology Co., Ltd		
		Beijing Psaitong Biotechnology Co., Ltd		
	Address	Building 145, Yougu New Science Park, Qingguang Town, Beichen District, Tianjin City		
	Tel/Fax	+86-10-60605840		
.5	Emergency phone number			
	Free ways ways a barrier ways have	+86-10-60605840		
	Emergency phone number	00-10-00003040		
	Emergency phone number Service hours	Monday to Friday, 9am-5pm (Standard time zone: UTC/GMT +8 hours).		
2.				
	Service hours	Monday to Friday, 9am-5pm (Standard time zone: UTC/GMT +8 hours).		
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no data available

### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Calcium carbonate	Calcium carbonate	471-34-1	207-439-9	100%

# 4. First-aid measures

### 4.1 Description of necessary first-aid measures

#### General advice

Medical attention is required. Consult a doctor. Show this safety data sheet (SDS) to the doctor in attendance.

#### If inhaled

Fresh air.

#### Following skin contact

Rinse skin with plenty of water or shower.

#### Following eye contact

Rinse with plenty of water (remove contact lenses if easily possible).

#### Following ingestion

Rinse mouth.

### 4.2 Most important symptoms/effects, acute and delayed

Exposure Routes: inhalation, skin and/or eye contact Symptoms: Irritation eyes, skin, respiratory system; cough Target Organs: Eyes, skin, respiratory system (NIOSH, 2016)

Exposure Routes: inhalation, skin and/or eye contact Symptoms: Irritation eyes, skin, mucous membrane; cough, sneezing, rhinorrhea (discharge of thin mucus); lacrimation (discharge of tears) Target Organs: Eyes, skin, respiratory system (NIOSH, 2016)

Exposure Routes: inhalation, skin and/or eye contact Symptoms: Irritation eyes, skin, mucous membrane, upper respiratory system; cough, sneezing, rhinorrhea (discharge of thin mucus); lacrimation (discharge of tears) Target Organs: Eyes, skin, respiratory system (NIOSH, 2016)

# 4.3 Indication of immediate medical attention and special treatment needed, if necessary

A serum calcium concentration exceeding 2.6 mmol per liter (10.5 mg per 100 mL) is considered a hypercalcemic condition. Withholding additional administration of calcium and any other medications that may cause hypercalcemia usually resolves mild hypercalcemia in asymptomatic patients, when patient renal function is adequate. Calcium supplements

# 5. Fire-fighting measures

# 5.1 Extinguishing media

### Suitable extinguishing media

In case of fire in the surroundings, use appropriate extinguishing media.

# 5.2 Specific hazards arising from the chemical

Not combustible.

# 5.3 Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media.

# 6. Accidental release measures

# 6.1 Personal precautions, protective equipment and emergency procedures

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Sweep spilled substance into covered containers.

### 6.2 Environmental precautions

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Sweep spilled substance

into covered containers.

### 6.3 Methods and materials for containment and cleaning up

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Sweep spilled substance into covered containers.

# 7. Handling and storage

# 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

# 7.2 Conditions for safe storage, including any incompatibilities

Separated from acids, aluminium, ammonium salts, fluorine and magnesium.Separated from acids, aluminium and ammonium salts.

# 8. Exposure controls/personal protection

### 8.1 Control parameters

### Occupational Exposure limit values

Component	Calcium carbonate				
CAS No.	471-34-1				
	Limit v	Limit value - Eight hours		- Short term	
	ppm	mg/m <sup>3</sup>	ppm	mg/m <sup>3</sup>	
Australia		10 (1)			
Canada - Québec		10			
France		10 inhalable aerosol			
Hungary		10 inhalable aerosol			
Ireland		10 (1)			
		4 (2)			
Latvia		6			
New Zealand		10 (1)			
Poland		10			
Singapore		10 (limestone, marble)			
Switzerland		3 respirable aerosol			
USA - OSHA		15 total dust			
		5 respirable dust			
United Kingdom		10 inhalable aerosol			
		4 respirable aerosol			
	Remar	ks			
Australia	(1) This value is for inhalable dust containing no asbestos and				
Ireland	(1) lnha	alable fraction (2) Respirable fraction			
New Zealand	(1) The value for inhalable dust containing no asbestos and less than 1% free silica.				

# 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### 8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear safety goggles.

Skin protection

Protective gloves.

#### **Respiratory protection**

Avoid inhalation of dust. Use local exhaust.

#### Thermal hazards

no data available

# 9. Physical and chemical properties

Physical state	Solid. Powder.
Colour	White.
Odour	Odorless
Melting point/ freezing point	825 °C. Remarks:Calcium carbonate (aragonite).;1 330 °C. Remarks:Calcium carbonate
	(calcite).
Boiling point or initial boiling poin	t 333.6⁰C at 760mmHg
and boiling range	
Flammability	Noncombustible Solid
Lower and upper explosion limit /	no data available
flammability limit	
Flash point	197°C
Auto-ignition temperature	no data available
Decomposition temperature	825°C
рН	pH = 8 to 9
Kinematic viscosity	no data available
Solubility	0.001 % (NIOSH, 2016)
Partition coefficient n-	no data available
octanol/water	
Vapour pressure	0 mm Hg (approx) (NIOSH, 2016)
Density and/or relative density	2.93 g/cm <sup>3</sup> . Temperature:20 °C.;2.71 g/cm <sup>3</sup> . Temperature:20 °C.
Relative vapour density	no data available
Particle characteristics	no data available

# 10. Stability and reactivity

### 10.1 Reactivity

Decomposes above 825°C . This produces corrosive fumes of calcium oxide. Reacts with acids, aluminium, ammonium salts, fluorine and magnesium.

### 10.2 Chemical stability

Indefinite shelflife.

### 10.3 Possibility of hazardous reactions

Not combustible.CALCIUM CARBONATE is non-combustible. Decomposes at high temperature (825°C) to give gaseous carbon dioxide and calcium oxide (quicklime). Incompatible with acids, alum, ammonium salts, fluorine, magnesium. Reacts with acids and acidic salts to generate gaseous carbon dioxide with effervescence (bubbling). The reaction with concentrated solutions of acids is rapid and exothermic. The effervesence can create extensive foaming. Ignites on contact with fluorine.

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Calcium carbonate ... ignite and burn fiercely in contact with fluorine. Fluorine: metal salts

### **10.6 Hazardous decomposition products**

When heated to decomposition it emits acrid smoke and irritating vapors.

# 11. Toxicological information

#### Acute toxicity

- Oral: LD50 Mouse oral 6450 mg/kg bw
- Inhalation: LC50 rat (male/female) > 3 mg/L air (analytical).
- Dermal: LD50 rat (male/female) > 2 000 mg/kg bw.

#### Skin corrosion/irritation

no data available

#### Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

no data available

#### **Reproductive toxicity**

no data available

#### STOT-single exposure

May cause mechanical irritation to the respiratory tract and eyes.

#### STOT-repeated exposure

Health effects of the substance have been investigated but none have been found

#### Aspiration hazard

A nuisance-causing concentration of airborne particles can be reached quickly when dispersed, especially if powdered.

# 12. Ecological information

### 12.1 Toxicity

- Toxicity to fish: LC50 Oncorhynchus mykiss (previous name: Salmo gairdneri) > 100 % v/v saturated solution 96 h.
- Toxicity to daphnia and other aquatic invertebrates: EC50 Daphnia magna > 100 % v/v saturated solution 48 h.
- Toxicity to algae: EC50 Desmodesmus subspicatus (previous name: Scenedesmus subspicatus) > 14 mg/L 72 h.
- Toxicity to microorganisms: EC50 activated sludge of a predominantly domestic sewage > 1 000 mg/L 3 h. Remarks:Respiration rate.

### 12.2 Persistence and degradability

no data available

#### 12.3 Bioaccumulative potential

no data available

### 12.4 Mobility in soil

no data available

### 12.5 Other adverse effects

no data available

# 13. Disposal considerations

13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

#### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# 14. Transport information

# 14.1 UN Number

	ADR/RID: Not dangerous goods. (For reference only, please check.)	IMDG: Not dangerous goods. (For reference only, please check.)	IATA: Not dangerous goods. (For reference only, please check.)
14.2	UN Proper Shipping Name		
	ADR/RID: Not dangerous goods. (For reference only, please check.)	IMDG: Not dangerous goods. (For reference only, please check.)	IATA: Not dangerous goods. (For reference only, please check.)
14.3	Transport hazard class(es)		
	ADR/RID: Not dangerous goods. (For reference only, please check.)	IMDG: Not dangerous goods. (For reference only, please check.)	IATA: Not dangerous goods. (For reference only, please check.)
14.4	Packing group, if applicable		
	ADR/RID: Not dangerous goods. (For reference only, please check.)	IMDG: Not dangerous goods. (For reference only, please check.)	IATA: Not dangerous goods. (For reference only, please check.)
14.5	Environmental hazards		
	ADR/RID: No	IMDG: No	IATA: No
14.6	Special precautions for user		
	no data available		

# 14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

no data available

# 15. Regulatory information

# 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Calcium carbonate	Calcium carbonate	471-34-1	207-439-9
European Inventory of Existing Commercial Chemical Substances (EINECS)			
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Not Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.
Korea Existing Chemicals List (KECL)			Listed.

# 16. Other information

Information on revision

Creation Date	July 15, 2024
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#### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

#### References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemlDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

#### Other Information

Calcium carbonate exists in nature as mineral aragonite and calcite (as in limestone, chalk and marble).

Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any damage resulting from handling or from contact with the above product.